

Synthesis and characterization of polychlorinated naphthalenes. VI. PCNs 39, 41, 60, 61 and 62

Vladimir A Nikiforov¹

¹St.petersburg State University, Dept. Of Chemistry

Introduction

In a series of presentations at DIOXIN symposia we have reported synthesis and characterization of a large number of polychloronaphthalenes¹⁻⁵. By the end of the last century only 10 congeners remained unknown or inadequately characterized – PCNs 2, 29, 39, 41, 45, 51, 58, 60, 61, 62. In particular, no melting point was reported for these 10 congeners, except PCN 2. This congener (2-chloronaphthalene) is the only liquid at ambient temperatures, with reported melting point from -17 °C to -2.3°C.

Our research is directed towards synthesis of all 75 PCN congeners and present work describes synthesis and properties of the following congeners

PCN 39 – 1,2,6,7-TeCN

PCN 41 – 1,2,7,8-TeCN

PCN 60 – 1,2,4,6,7-PeCN

PCN 61 – 1,2,4,6,8-PeCN

PCN 62 – 1,2,4,7,8-PeCN

Materials and Methods

Starting PCN congeners were previously synthesized in our laboratory¹⁻⁶. Nitration was carried out with equivalent amount of NO₂BF₄ in sulfolane at ambient temperature^{1,4}. Chlorosulfonation was carried out in excess of neat HSO₃Cl. Substitution of nitro- and chlorosulfonyl groups was achieved in boiling C₅Cl₆^{4,7}. For hydrodechlorination Cu in boiling acetic acid was employed^{5,6}.

Nitro-PCNs and PCN-sulfonylchlorides were purified by crystallization from ethanol. PCNs were purified by chromatography on silicagel column with hexane as eluent; then crystallized from MeOH/CH₂Cl₂. All chemicals and reagents were used as received.

Course of reactions, separations and purifications and purities of products were controlled by GC/ECD. GC conditions were as follows : GC-Varian3700, inj. – Gerstel split/splitless at 250°C, column – DB-5(app. 50m), Det. – ECD(at 300°C), carrier gas – nitrogen, make-up – nitrogen. Pr.: 160 °C(2 min) - 20 °C/min - 280 °C(10 min) Purge 1.00-1.90 min. Structures were established by NMR.

Results and Discussion

Structures of “missing” PCN congeners are given below (Fig. 1). Interestingly, one mono-CN, four tetraCNs and 5 penta-CNs have not been either reported at all or incompletely characterized by the beginning of our century.

EMG - Polychlorinated Naphthalenes

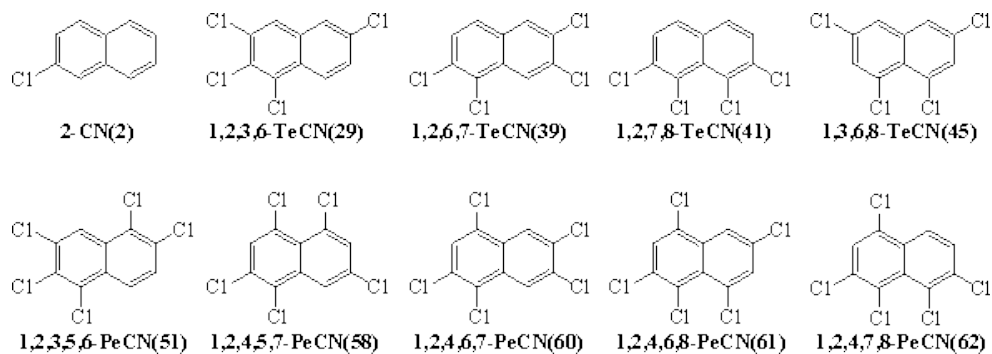
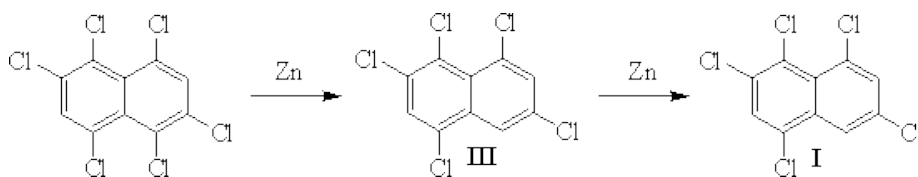


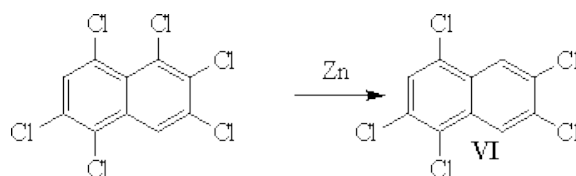
Figure 1. Structures of unknown or incompletely characterized PCN congeners. In brackets – PCN numbers.

Five of the ten missing congeners were prepared according to the scheme on Fig. 2.

For confirmation purposes 1,2,4,5,7-PeCN(III) was also obtained by alternative synthetic route - as intermediate by reduction of 1,2,4,5,7,8-HxCN with Zn^{5,6}.



1,2,4,6,7-PeCN(VI) was similarly prepared in minuscule amount from 1,2,3,5,7,8-HxCN⁶.



Nitration of 1,2,4,7-TeCN(IV) gave a mixture of products, two were isolated in pure state by chromatography on silicagel with hexane as eluent – 3-nitro-1,2,4,7-TeCN(VII) and 8-nitro-1,2,4,7-TeCN(VIII). The latter was converted to 1,2,4,7,8-PeCN(IX). This compound was also obtained by alternative route XIII → X → XI → XII → IX.

Hydrodechlorination of 1,2,3,7,8-PeCN(XIV) with Zn yields target 1,2,6,7-TeCN(XV), ca. 80% and side product 1,2,3,7-TeCN, ca 20%^{5,6}. The mixture was separated on silicagel with hexane as eluent.

Melting points, GC retention times and ¹H NMR chemical shifts are summarized in Table 1.

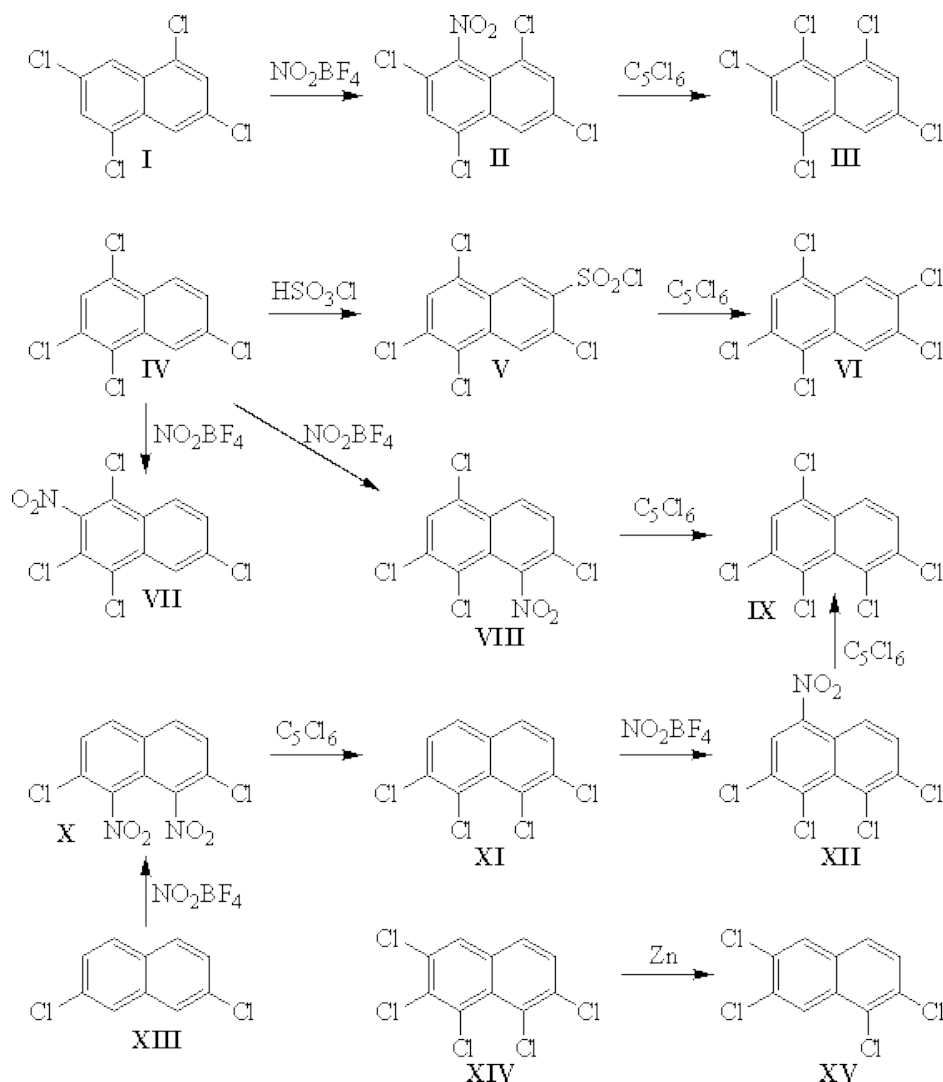


Figure 2. Synthesis of 1,2,6,7-TeCN(XV), 1,2,7,8-TeCN(XI), 1,2,4,6,7-PeCN(VI), 1,2,4,6,8-PeCN(III) and 1,2,4,7,8-PeCN(IX).

Table 1. Melting points, GC retention times(RT) and ^1H NMR chemical shifts of PCNs and related compounds (compound numbers – Fig. 2)

Compound number	RT	Melting point	^1H NMR chemical shifts, d(ppm)						
			H-2	H-3	H-4	H-5	H-6	H-7	H-8
I	6.52,2	176	7.63d	-	8.14d	-	7.63d	-	8.14d
III	8.30,4	155.5	-	4.76s	-	8.26d	-	7.74d	-
IV	7.08,3	143	-	7.64s	-	8.18d	7.59dd	-	8.29d
V	-	-	-	7.83s	-	9.10s	-	-	8.56s
VI	8.18,1	130.2	-	7.66s	-	8.41s	-	-	8.35s
VII	-	-	-	-	-	8.31d	7.74dd	-	8.36d
VIII	9.32,8	109	-	7.83s	-	8.41d	7.71d	-	-
IX	8.49,6	103.4	-	7.75s	-	8.20d	7.69d	-	-
X	10.26,8	256	-	7.75d	8.05d	8.05d	7.75d	-	-
XI	8.07,2	128.1	-	7.60d	7.68d	7.68d	7.60d	-	-
XII	9.33,0	-	-	-	-	-	-	-	-

XIV	9.18,7	117	-	-	7.87s	7.59s	7.59s	-	-
XV	7.36,8	156.9	-	7.53d	7.61d	7.94s	-	-	8.39s

Thus we have half-finished the remaining work on synthesis and characterization of all 75 PCN congeners. Synthesis of 1,2,3,6-TeCN, 1,3,6,8-TeCN, 1,2,3,5,6-PeCN and 1,2,4,5,7-PeCN is in progress.

Acknowledgement

Prof. Jaakko Paasivirta is gratefully acknowledged for encouragement to this work.

References

1. Nikiforov V.A., Auger P., Wightman R.H., Malaiyandi M., Williams D.T. (1992) *Organohalogen Compounds*, 8: 123-124.
2. Nikiforov V.A., Karavan V.S., Miltsov S.A., Tribulovich V.G. (1993) *Organohalogen Compounds*, 14: 229-230.
3. Nikiforov V.A., Miltsov S.A., Karavan V.S., Tribulovich V.G., Vlasov S.V., Wightman R.H. (1994) *Organohalogen Compounds*, 19: 137-138.
4. Nikiforov V.A., Karavan V.S., Miltsov S.A., Tribulovich V.G. (1998) *Organohalogen Compounds*, 35: 159-162.
5. Vladimir A. Nikiforov, Sergei A. Miltsov, Vladimir S. Karavan and V.V. Varentsov (2000) *Organohalogen Compounds*, 47: 171-173.
6. S. Miltsov, V. Karavan, V. Nikiforov, V. Tribulovich, V. Varentsov (1999) *Zh. Org. Khim. (rus)*:724-727.
7. V. Nikiforov, R. Wightman (1997) *Chimia* 51: 452.