

MULTIPLE REACTION MONITORING CONDITIONS FOR THE SIMULTANEOUS DETERMINATION OF 2,3,7,8-SUBSTITUTED PCDD/Fs, PCBs 77, 81, 126, 169 AND THE PARALLEL ANALYSIS OF 23 PCBs IN FOODS USING MINI-BORE GAS CHROMATOGRAPHY AND TANDEM MASS SPECTROMETRY

Douglas Hayward

US Food and Drug Administration, 5100 Paint Branch Parkway, College Park, Maryland, 20740

Introduction

Although food levels have significantly fallen for PCDD/Fs during the past two decades, the demands for isomer specific PCDD/F and PCB determinations have increased. A major focus has been in the area of foods and animal feeds and their components. Three recent animal food contamination incidences have increased the interest in comprehensive monitoring¹⁻³ of food/feed and feed additives. Recent discoveries of feed additive contamination have sustained an interest in monitoring the most toxic PCDD/F and PCB congeners. FDA has recently greatly expanded the number and type of foods/feeds to be measured annually and will include 26 PCB congeners. The large number of foods, feeds, feed additives that require examination during the course of monitoring programs and contamination events prompted investigators to search for screening techniques. Field-tested screening techniques fall into two categories: bio-analytical assays⁴⁻⁸ or fast GC(MS)^{2,9} with high volume injection¹⁰ and/or coupled to cheap MS techniques^{2,10,12-14}. The screening methods require low false negative rates at the action levels while positives and some selected negatives are measured by HRGC/HRMS^{3,13}. Tandem mass spectrometry (TMS) in a quadrupole ion trap has been suggested for quick screening for dioxins for several years¹¹. Initial reports were limited to TCDD without uniformly labeled TCDD due to software limitations, but produced useful results in three matrices including fish. Recently, a number of reports have suggested useful applications to wastewater, sludge¹², food¹³ ash¹⁴, and as an alternative to immunoassay screening¹⁴.

Ion traps produce collision induced dissociation product spectra of stored ions through the application of a waveform(s) across the end-caps matching the periodic motion of stored ion. Increasing the energy of a stored ion can increase the amplitude of the ion motion in the z-axis resulting in either ion ejection as is done in mass axial instability scanning for mass spectrum collection or ion dissociation. Conditions appropriate for each targeted chemical can maximize the formation of product ion with minimal ion loss. The purpose of these studies were to select conditions that optimize a single dissociation channel for PCDDs and PCDFs (M-COCl) and for polychlorobiphenyls (M-Cl₂) and then evaluate these conditions through analysis of first finfish and then other foods.

Methods

Method optimization was performed on a Saturn 2000 quadrupole ion trap equipped with septum programmable on column injection through a Merlin Microseal septumless injector cap. PCDD/Fs and/or PCBs were separated on a 40 M mini-bore (0.18 mm ID). Temperature program for PCDD/Fs was 140 °C (2 min. hold) to 200 °C at a rate of 20 °C/min, then to 240° (12 min hold) at a rate of 5 °C/min, and finally to 280 °C at a rate of 10°/min. PCB temperature program was 120 °C (2 min hold)

to 200 °C at a rate of 20 °C/min to 220 °C (10 min. hold) at a rate of 2 °C/min, and finally to 290 °C at a rate of 10°/min.

PCB product ion spectra were adjusted to the highest response while maximizing the spectral difference between mon-ortho chlorine substituted congeners and di-ortho chlorine substituted congeners. Twenty-six instrument parameters for PCDD/F and PCBs are optimized through tests with all targeted analytes¹⁵. The final parameter conditions for the 44 congeners are evaluated with a five point standard curve and initially through the analysis finfish species collected in Lake Chardara, Kazakhstan and bluefish, Atlantic salmon, and rockfish obtained commercially in Washington, DC USA¹⁶. Test portions of finfish were fortified with 100 pg ¹³C₁₂-labeled PCDD/Fs, 250 pg ¹³C₁₂-labeled PCBs 77, 126, 169 and 1000 pg ¹³C₁₂-PCBs 28, 52, 101, 105, 114, 118, 138, 153, 156, 157, 167 180, 189. Fish samples were prepared as previously described¹⁷ except the PCB containing fractions were saved and clean up over a multi-layer K⁺silicate/sulfuric acid silica gel/neutral silica gel column followed with Florisil or a neutral alumina column.

Results and Discussion

Conditions selected for PCDD/Fs and PCBs 77, 81, 126, and 169 are listed in Table 1. PCDD/Fs all use M+1 for the precursor ion, except for HpCDD/Fs and OCDD/F (M+3), while PCB precursor selection utilizes M+2. PCDD/Fs are dissociated using a narrow band of frequencies (5-9) so that the stored precursor ions, M+ and M+2, are completely dissociated. The resulting product ion spectrum contains product ions with chlorine isotopic responses that are of equal or nearly equal intensity when part of the loss contains a single chlorine atom (M-COCl, M-Cl, M-CO₂Cl etc). Instrumental LODs for TCDD are approximately 100 fg for either product ion using these GC conditions¹⁵. PCDDs and PCDFs were standardized from 0.4 – 200 pg or 4-1000 pg for PCBs 77, 126 and 169 with an average RSD for all 20 congeners of 6%. The standard curve for the remaining PCBs between 5-5000 pg gave an average RSD of 11% for all congeners. LODs for PCBs 77, 126 and 169 were found to be similar to the PCDD/Fs. Repeat analysis of bluefish, rockfish and salmon for the detected PCDD/Fs and PCBs 77,126 and 169 gave RSDs below the 20% method criteria. Figure 1 shows the product ion spectra for a mon-ortho PCB and di-ortho PCB dissociated using a single frequency at 2V excitation amplitude. Notice that the presence of a PCB-99 (di-ortho-pentachloro congener) is easier to detect due to a larger response from M-Cl when excitation energy is set at 2V rather than at 4V and is nearly absent from PCB 118 under all excitation energies. Because mono-ortho congeners have greater dioxin-like toxicity, it is important to distinguish them from potentially co-eluting di-ortho congeners.

Conclusions

Product ion response for PCBs were higher using a single frequency during excitation rather than a narrow band as is used for the PCDD/Fs. PCDD/F conditions produced high sensitivity while also monitoring non-ortho-PCBs. Because the selected product ion loss for dioxin-like PCBs is the loss of 2 chlorine atoms, no advantage is found by fragmenting precursor ions other than M+2 (or M+4). The spectral differences found with the lower excitation energies provide a high degree of selectivity, while maintaining sufficient sensitivity. The spectral differences also allow a constant check on chromatographic resolution of closely eluting mono-ortho and di-ortho congeners, thereby reducing or eliminating the need for extensive separations before the GC/MS.

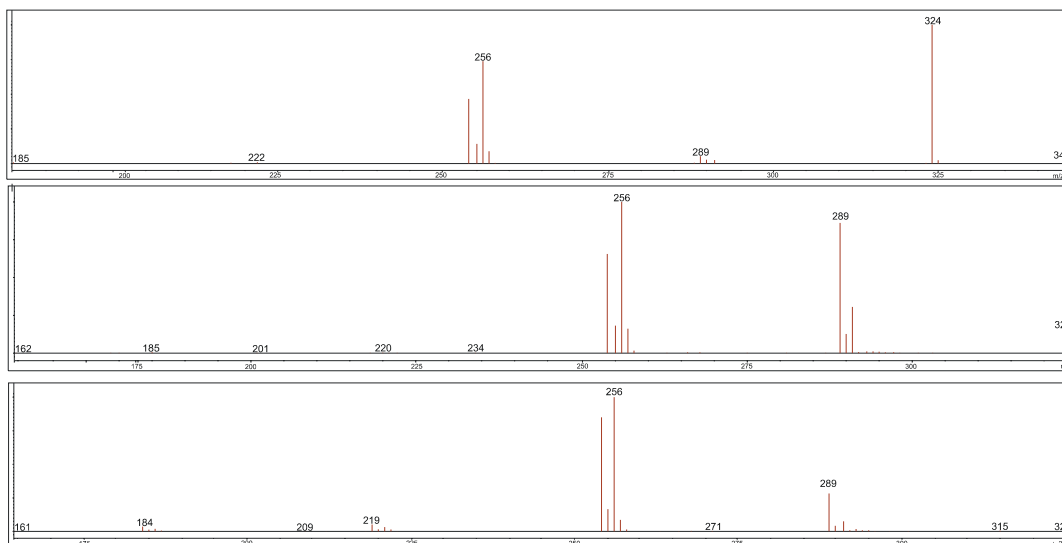


Figure 1: Product spectra; PCB-118, PCB-99 at 2V top/middle respectively; PCB-99 (bottom) at 4V.

References

1. Ferrario J. B., Bryne C. J. and Cleverly D. H. (2000) *Environ. Sci. Technol.* 34, 4524.
2. Hayward D.G., Nortrup D., Gardner A., and Clower Jr. M. (1999) *Environ. Res.* 81, 248.
3. Van Overmeire I., Goeyens L., Beernaert H., Srebrnik S., De Poorter G., Baeyens W., Clark G. Chu M. Chu A., Morris R. and Brown D. (2000) *Organohalogen Compounds* 45, 196.
4. Sugawara Y. Gee S. J., Sanborn J. R., Gilman S.D. and Hammock B.D. (1998) *Anal. Chem.* 70, 1092.
5. Chiu Y.-W., Carlson R. E.; Maucus K. L. and Karu A.E. (1995) *Anal. Chem.* 67, 3829.
6. Bosveld A. T. C., Vanden Berg M. (1994) *Fresenius Journal of Anal. Chem.* 348, 106.
7. Garrison P. M., Tullis K. Aarts J. M. M. J. G., Brouwer A., Giesy J. P. and Denison M. S. (1996) *Fund. App. Toxicol.* 30, 194.
8. Bovee T. F. H. Hoogenboom L. A. P., Traag T. Z., Aarts J. M. M. J. G., Brouwer A. and Kuiper H. A. (1998) *Food Add. Contam.* 15, 863.
9. MacPherson K.A., Reiner E.J. and Kolic T.M. (2001) *Organohalogen Compounds* 50, 40.
10. Epe G., Focant J.-F., Pirard C. and de Pauw E. (2001) *Organohalogen Compounds* 50, 186.
11. Plomley J. B., Koester C. J. and March R. E. (1994) *Organic Mass Spect.* 29, 372.
12. Kuchler T. and Brzezinski H. (2000) *Chemosphere* 40, 213.
13. Hayward D.G., Holcomb J., Glidden R., Wilson P., Harris M. and Spencer V. (2001) *Chemosphere*, 43, 407.
14. Focant J.-F., Epe G. and De Pauw E. (2001) *Chemosphere* 43, 417
15. Hayward D.G. (2002) *Organohalogen Compounds* 55, 99.
16. Hayward D.G. and Hooper, K. (2003) submitted to *Organohalogen Compounds*.
17. Hayward D.G., Hooper K. and Andrzejewski D. (1999) *Anal. Chem.* 71, 212.

Table 1: Quadrupole ion storage tandem mass spectrometry (QISTMS) conditions¹; PCDD/F and PCB acquisition segments in elution order on a DB-5 ms capillary column.

Segment (#)/ Congener	Precursor ion (m/z)	LMCO (m/z)	CID (V)	CID freq. #	Product ion m/z	TIR ² (Obs.)
(1) PCB 77+81	292	129	2.0	1	220,222	1.0(1.04)
	¹³ C ₁₂	304	135	2.0	1	
(2) TCDF	305	135	3.5	9	241,243	1.37(1.6)
	¹³ C ₁₂	317	140	3.4	9	
(3) TCDD	¹³ C ₁₂ -TCDD	334	148	3.0	9	268, 270
	321	141	2.8	9	257, 259	1.37(1.4)
PCB-126	¹³ C ₁₂	336	148	3.0	9	268, 270
	326	144	2.25	1	254, 256	0.66(0.64)
(4) PeCDF	¹³ C ₁₂ -PCB 126				266, 268	
	339	151	3.8	9	275, 277	1.03(1.15)
PeCDD	¹³ C ₁₂	351	156	3.8	9	286, 288
	355	155	2.0	5	291, 293	1.03(1.07)
PCB-169	¹³ C ₁₂	367	162	2.0	5	302, 304
	360	159	2.5	1	288, 290	0.5(0.49)
(5) HxCDF	¹³ C ₁₂ -PCB-169	372	164	2.5	1	300, 302
	373	166	4.0	9	309, 311	0.82(0.84)
HxCDD	¹³ C ₁₂	385	171	4.0	9	320, 322
	389	171	3.0	13	325, 327	0.82(0.86)
(6) HpCDF	¹³ C ₁₂	401	178	3.0	13	336, 338
	409	181	3.6	13	345, 347	1.86(2.03)
HpCDD	¹³ C ₁₂	421	187	3.5	13	356, 358
	425	187	2.0	5	361, 363	1.86(1.69)
(7) OCDF	¹³ C ₁₂	435	193	2.0	5	370, 372
	444	194	4.0	9	379, 381	1.53(1.4)
OCDD	¹³ C ₁₂	460	201	3.5	9	395, 397
	471	209	3.3	9	406, 408	1.53(1.6)

¹CID time (ms) = 10 (PCBs, 4 ms), mass window = 3 m/z (334=1 m/z), maximum ion time = 65 msec, emission current = 100 μ amps, scan rate = 0.27 sec, multiplier offset = +300 volts, target RIC = 5,000. LMCO = low-mass cut off (CID). Mass window ¹³C₁₂-2,3,7,8-TCDD = 5 and 2 for PCBs and low and high mass offsets set to 2 DACs. ²TIR(Obs.) = theoretical isotope ratio(observed).