

ANALYSIS

Calculated NMR ^{13}C Spectra of Polyhalogenated Biphenyls (Cl and Br)

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1. Introduction

As is known ^{13}C NMR chemical shifts in polysubstituted compounds are related with molecular structures by a limited number of parameters, so it enables to reconstruct NMR spectra of all species of interest¹⁾. Earlier we created the method of simulating ^{13}C NMR spectra of organohalogen compounds with mono- and two-particle increment scheme which was used for prediction of ^{13}C NMR spectra of wide group of polysubstituted organohalogen aromatics, include polychlorinated (PCB)²⁾ and polybrominated biphenyls (PBB)³⁾. Data of this research and also of mixed substituted polyhalogenated benzenes⁴⁾ allow us to predict ^{13}C NMR for biphenyls, containing Cl and Br simultaneously (PCBBs).

2. Discussion

Tables 1 and 2 show some ^{13}C NMR spectra for respectively four- and five-substituted PCBBs of two types: high-toxic compounds with halogens in 3,4,5,3',4'-positions, and low-toxic biphenyls, with halogen in *ortho*-position. Fig.1 demonstrates changes, occurring in spectra of two pairs of five-substituted PCBBs when *para*-halogen passed to *ortho*-position. The spectra find out essential distinctions, thus, the solution of a 4-Hal or 5-Hal and other congener mixture can be investigated by various methods of 2D- ^{13}C NMR spectroscopy for recognition of signals, belonging to one substance, and then to identify each compound by comparison of experimental and calculated data.

3. References

- 1) Fedorov L.A. and V.I.Dostovalova. NMR approach to structure analysis of aromatic toxicants in environment. *Chemosphere* 1995; **30**(7): 1311-1329.
- 2) Dostovalova V.I. and L.A.Fedorov. The elucidation of chlorinated biphenyls structures by prediction of NMR ^{13}C chemical shifts. *Organohalogen compd.* 1994; **19**, 195-197.
- 3) Dostovalova V.I. and Fedorov L.A. Elucidation of brominated Biphenyl structures in the framework of the common method for prediction of NMR ^{13}C chemical shifts for polysubstituted benzenes. *Organohalogen compd.* 1995; **26**, 431-433.
- 4) Fedorov L.A. and V.I.Dostovalova. The prediction of NMR ^{13}C chemical shifts for Mixed Cl,Br-benzenes in frame of two-particle Increments. *Organohalogen compd.* 1996; **28**, 456-458.

Dioxin '97, Indianapolis, Indiana, USA

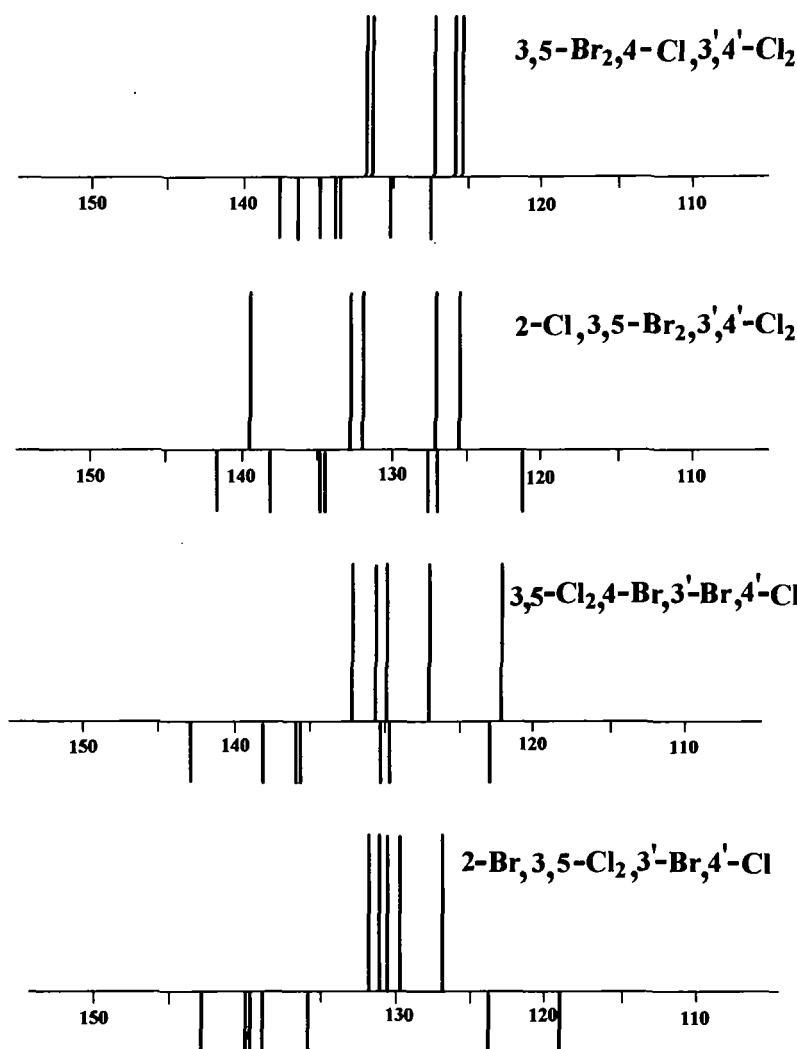


Fig. 1

^{13}C NMR spectra calculated for two pairs of five-substituted PCBBs when *para*-halogen passed to *ortho*-position (Cl for the 1-st pair and Br for the 2-nd.). The spectra are shown in TMS scale, in a mode of J-modulated spin echo, to distinguish C-atoms with hydrogen and without it.

Table 1. The calculated ^{13}C NMR chemical shifts for some high- and low-toxic tetrasubstituted PCBBs

	1	2	3	4	5	6	1'	2'	3'	4'	5'	6'
nil	140.6	126.75	128.37	126.87	128.37	126.75	140.6	126.75	128.37	126.87	128.37	126.75
toxic PCBB												
3-Cl,4-Br,3',4'-Br₂	144.29	126.76	136.75	131.18	125.02	128.86	144.04	129.25	125.18	135.04	123.97	128.04
3,4-Cl₂ 3',4'-Br₂	138.68	129.89	133.75	133.19	131.3	126.52	142.7	127.04	123.62	132.51	125.15	128.58
3-Br,4-Cl, 3',4'-Br₂,	139.14	132.01	123.80	139.16	131.3	127.51	144.4	127.18	125.3	135.47	124.09	127.97
3-Cl,4-Br, 3'-Br,4'-Cl	143.72	124.09	136.26	130.49	124.53	128.19	138.21	131.41	123.19	138.04	130.69	126.91
3-Cl,4-Br,3'-Cl,4'-Br	143.36	126.16	136.14	130.06	124.41	128.26	143.36	126.16	136.14	130.06	124.41	128.26
3-Cl,4-Br,3',4'-Cl₂	142.29	123.64	136.18	130.1	124.25	127.74	136.15	130.67	137.88	127.91	133.45	127.87
3-Br,4-Cl,3',4'-Cl₂	134.03	126.75	131.9	127.21	133.14	129.24	138.51	127.02	133.26	132.3	130.61	125.65
3-Br,4-Cl,3'-Br,4'-Cl	139.26	128.65	123.51	137.03	132.05	126.84	140.5	129.34	123.51	136.54	130.81	126.64
mono-ortho-Hal PCBB												
2-Br,3-Cl,3',4'-Br₂	142.06	123.66	139.45	131.41	133.22	129.76	140.37	133.41	126.31	124.3	135.14	128.25
2,3-Cl₂,3',4'-Br₂	139.42	131.6	136.15	130.19	128.2	127.8	139.02	133.25	125.76	123.68	134.59	128.09
2-Cl,3-Br3',4'-Br₂	141.04	135.5	126.4	132.25	128.97	128.79	139.95	133.85	126.37	124.8	135.2	128.69
2-Br,3-Cl,3'-Br,4'-Cl	141.09	121.19	138.96	130.92	132.93	129.29	141.54	131.91	123.89	135.91	131.19	127.21
2-Br,3-Cl, 3'-Cl,4'-Br	141.13	123.06	138.84	130.29	132.61	129.16	139.22	131.3	138.14	121.85	133.22	127.26
2-Br,3-Cl,3',4'-Cl₂	140.16	120.59	138.35	129.8	132.32	128.69	139.15	129.79	133.64	131.87	131.19	126.42
2-Cl,3-Br,3',4'-Cl₂	138.94	135.86	125.3	136.26	128.07	129.15	138.93	126.8	133.7	126.75	131.25	125.43
2-Cl,3-Br,3'-Br,4'-Cl	140.07	133.03	125.91	131.76	128.68	128.32	141.12	132.35	123.95	136.41	131.25	127.65

Table 2. The calculated ^{13}C NMR chemical shifts for some high- and low-toxic pentasubstituted PCBBs

	1	2	3	4	5	6	1'	2'	3'	4'	5'	6'
• nil	140.6	126.75	128.37	126.87	128.37	126.75	140.6	126.75	128.37	126.87	128.37	126.75
toxic PCBB												
3,5-Cl₂,4-Br,3',4'-Br₂,	145.84	124.72	137.12	130.99	130.31	130.66	142.21	129.0	125.1	134.85	123.89	127.79
3,4-Cl₂,5-Br,3',4'-Br₂	141.38	128.84	136.82	136.32	125.83	129.28	141.8	127.39	124.15	133.44	125.68	128.93
3,5-Br₂,4-Cl,3',4'-Br₂	141.84	130.96	126.87	142.29	125.83	130.27	143.5	127.53	125.83	136.4	124.62	128.32
3,5-Cl₂,4-Br,3'-Br,4'-Cl	145.27	122.05	136.63	130.3	129.82	129.99	136.38	131.16	123.11	137.85	130.61	126.66
3-Cl,4,5-Br,3'-Cl,4'-Br	146.06	125.11	138.43	132.32	117.87	132.17	142.46	126.51	136.67	130.99	124.94	128.61
3,5-Cl,4-Br,3',4'-Cl₂	143.84	121.60	136.55	129.91	129.54	129.54	134.32	130.42	137.8	127.72	133.37	127.62
3,5-Br,4-Cl,3',4'-Cl₂	136.73	125.7	134.97	130.33	127.67	132.0	137.61	127.37	133.79	133.23	131.14	126.0
3-Br,4,5-Cl,3'-Br,4'-Cl	140.81	126.61	125.34	135.63	136.33	128.17	138.67	129.09	123.43	136.35	130.73	126.39
mono-ortho-Hal PCBB												
2-Br,3,5-Cl,3',4'-Br₂	143.61	121.62	140.51	131.22	138.71	130.32	138.54	133.16	126.23	124.11	135.06	128.0
2,3-Cl₂,5-Br,3',4'-Br₂	142.12	130.55	137.98	133.52	122.73	131.25	138.12	133.6	126.29	124.61	135.12	128.44
2-Cl,3,5-Br,3',4'-Br₂	143.74	134.45	128.23	135.58	123.5	132.24	139.05	134.2	126.9	125.73	135.73	129.04
2-Br,3,5-Cl,3'-Br,4'-Cl	142.64	119.15	140.02	130.73	138.42	129.85	139.71	131.66	123.81	135.72	131.11	126.96
2,5-Br,3-Cl,3'-Cl,4'-Br	143.83	122.01	140.67	133.62	127.14	132.61	138.32	131.65	138.67	122.78	133.75	127.61
2-Br,3,5-Cl,3',4'-Cl₂	141.71	118.55	139.41	129.61	137.81	129.25	137.32	129.54	133.56	131.68	131.11	126.17
2-Cl,3,5-Br,3',4'-Cl₂	141.64	134.81	127.13	139.59	122.6	132.6	138.03	127.15	134.23	127.68	131.78	125.78
2,5-Cl,3-Br,3'-Br,4'-Cl	141.62	130.99	126.97	131.57	134.17	128.88	139.29	132.10	123.87	136.22	131.17	127.40