

SPECTROPHOTOMETRIC MULTICOMPONENT ANALYSIS APPLIED TO PRIORITY POLLUTANT  
CHLOROPHENOLS. STANDARD SPECTRA BY MULTIVARIATE CALIBRATION.

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ABSTRACT

A multicomponent analysis method is proposed for the determination of chlorophenols, applying diode array derivative spectrophotometry. The method is based on formation of ion-pairs with tetrabutylammonium nitrate as counterion and extraction in chloroform at pH 9.1. Prediction and quantitative determination (concentration range 3-15 mg l<sup>-1</sup>) of two, three and four component mixtures were possible using multivariate calibration of first derivative spectra.

INTRODUCTION

Chlorophenols are widely used in pesticide formulations. The presence of chlorophenols in water, soil, food and animal feed stuffs presents a potential hazard owing to their toxicity. Spectrophotometric determination of these phenols is of limited use because of overlapping of their absorption bands (1). The need for reliable determination of chlorophenols has led to the development of a number of sophisticated analytical methods (2-5) mainly using chromatographic techniques (HPLC and GC). However, in laboratories where large number of samples must be processed rapidly a method with sufficient sensitivity and selectivity, which can be used for preliminary screening, has obvious advantages.

This paper describes the development of a method for the determination of chlorophenols using diode array derivative spectrophotometry. The method is based on formation of ion-pairs with tetrabutylammonium nitrate as counterion and extraction in chloroform at pH 9.1.

Prediction and quantitative determination of two, three and four component mixtures was possible using multivariate calibration (6) of first derivative spectra and multicomponent analysis computer programme based on least squares linear regression.

## EXPERIMENTAL PART

### Reagents

Pentachlorophenol (99% pure, Carlo Erba) o-chlorophenol (99% pure, Carlo Erba), 2,4-dichlorophenol (99% pure, Aldrich-Chemie), 4-chloro-3 methyl phenol (99% PURE? Aldrich-Chemie) and 2,4,6-trichlorophenol (98% pure, Carlo Erba) were used for preparing 150 and 50  $\text{mg l}^{-1}$  stock solutions. Standard working solutions of the chlorophenols were prepared by suitable dilution of the stock solutions with water. A 5000  $\text{mg l}^{-1}$  stock solution of tetrabutylammonium nitrate (98% pure, Fluka Chemie AG) was also used.

The pH was adjusted with a 0.1M phosphate buffer (pH 9.1). Spectrophotometric grade chloroform (Carlo Erba) was used for extraction. All chemicals used were analytical grade and water was obtained using a Milli Q apparatus.

### Apparatus

A detection system consisted of a diode array HP 8452 A spectrophotometer equipped with 4 cm path length silica cell. The spectrophotometer was interfaced to an HP Vectra AI computer and a HP Think Jet printer.

### Procedure

Aliquots of chlorophenol solutions containing two, three or four chlorophenols in the 3-15  $\text{mg l}^{-1}$  range were transferred into a 10 ml beaker and the following reagents were added: 3 ml of the tetrabutylammonium nitrate solution and 3 ml of the phosphate buffer solution; the volume was made up to the mark with water. This solution was treated with 10 ml of chloroform in a 100 ml extraction funnel and stirring for 3 min. The absorbance of the organic extract was measured between 280-400 nm against the blank.

## RESULTS AND DISCUSSION

The first derivative standard spectra of chlorophenols (2-chlorophenol, 4-chloro-3-methylphenol, 2,4-dichlorophenol and pentachlorophenol) has been obtained from twenty quaternary mixtures containing chlorophenols in concentrations between 3-15  $\text{mg l}^{-1}$ . Binary and ternary mixtures have been resolved using first derivative standard spectra of twenty binary and ternary mixtures containing chlorophenols in the concentration range 3-9  $\text{mg l}^{-1}$ . The multicomponent computer programme allows the determination of each phenol in such mixtures in the wavelength range 280-400 nm with relative errors below 5.4%. Results obtained from quaternary mixtures are shown in the table. As it can be seen the overlapping of ionic pair spectra in the range 280-400 nm does not prevent simultaneous determination of the four chlorophenols whose errors is acceptable for particular application.

## MULTIVARIATE-MULTICOMPONENT DETERMINATION OF QUATERNARY CHLOROPHENOLS MIXTURES

Mixture composition, results and errors

,4-dichlorophenol, $\text{mg l}^{-1}$			4-chloro-3methyl phenol, $\text{mg l}^{-1}$			2,4-dichlorophenol, $\text{mg l}^{-1}$			pentachlorophenol, $\text{mg l}^{-1}$		
<u>Added*</u>	<u>Found</u>	<u>Error, %</u>	<u>Added*</u>	<u>Found</u>	<u>Error, %</u>	<u>Added*</u>	<u>Found</u>	<u>Error, %</u>	<u>Added*</u>	<u>Found</u>	<u>Error, %</u>
.0	4.9	1.4	5.0	5.0	0.0	9.0	9.4	4.8	7.0	7.2	2.3
.0	9.1	1.5	11.0	11.1	1.0	9.0	8.7	3.5	5.0	4.9	1.4
.0	9.3	3.1	11.0	11.3	2.9	13.0	13.1	0.3	7.0	7.1	1.4
5.0	15.6	4.0	15.0	15.2	1.7	9.0	9.2	2.1	7.0	7.2	2.6
3.0	13.2	1.5	9.0	9.0	0.0	9.0	9.2	2.1	5.0	4.9	1.0
.0	5.1	2.6	11.0	10.8	1.6	3.0	2.9	0.3	9.0	9.1	1.2
.0	6.8	2.7	7.0	6.8	3.0	9.0	9.3	3.6	9.0	8.9	1.3

Mean of three determinations.

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